## INTERNATIONAL JUNIOR SCIENCE OLYMPIAD Jakarta - Indonesia December 5-14,2004

## Solutions for TEST COMPETITION

1. The mobile system in the Figure is in the equilibrium condition. The object of $\mathrm{m}_{\mathrm{A}}$ has a mass of 0.5 kg and hang at the first crossbar. The second crossbar supports the mass of $\mathrm{m}_{\mathrm{B}}$ and $\mathrm{m}_{\mathrm{C}}$. Determine the tension F at the first crossbar and the masses of the objects $\mathrm{m}_{\mathrm{B}}$ and $\mathrm{m}_{\mathrm{C}}$ by neglecting the weights of crossbars. ( $\mathrm{g}=9.8 \mathrm{~m} / \mathrm{s}^{2}$ ).

A. $\quad \mathrm{F}=6.37 \mathrm{~N}, \mathrm{~m}_{\mathrm{B}}=0.12 \mathrm{~kg}, \mathrm{~m}_{\mathrm{C}}=0.03 \mathrm{~kg}$
B. $\mathrm{F}=5.37 \mathrm{~N}, \mathrm{~m}_{\mathrm{B}}=0.12 \mathrm{~kg}, \mathrm{~m}_{\mathrm{C}}=0.03 \mathrm{~kg}$
C. $\quad \mathrm{F}=6.37 \mathrm{~N}, \mathrm{~m}_{\mathrm{B}}=0.10 \mathrm{~kg}, \mathrm{~m}_{\mathrm{C}}=0.03 \mathrm{~kg}$
D. $F=6.37 \mathrm{~N}, \mathrm{~m}_{\mathrm{B}}=0.12 \mathrm{~kg}, \mathrm{~m}_{\mathrm{C}}=0.01 \mathrm{~kg}$

## Answer: A

## Solution

Look at the sub-system at the upper crossbar, and consider the center of rotation at the F.

$$
\begin{aligned}
\sum \tau & =0 \\
0 & =-0.03 \times 0.5 \mathrm{~kg} \times 9.8 \mathrm{~m} / \mathrm{s}^{2}+0.1 F_{C B} \\
F_{B C} & =\frac{0.03 \times 0.5 \cdot 9.8}{0.1}=1,47 \mathrm{~N}
\end{aligned}
$$

Look at the sub-system of $\mathrm{m}_{\mathrm{B}} \mathrm{m}_{\mathrm{C}}$, and consider the center of rotation at the edge of $\mathrm{m}_{\mathrm{B}}$.

$$
\begin{aligned}
\sum \tau & =0 \\
& =0.02 \cdot F_{B C}-0.11 \cdot F_{C} \\
F_{C}=\frac{0.02 \cdot F_{B C}}{0.11} & =\frac{0.02 \times 1.47 \mathrm{~N}}{0.11} \\
& =0,267 \mathrm{~N} \\
m_{C} & =\frac{F_{C}}{g}=\frac{0.267 \mathrm{~N}}{9.8 \mathrm{~m} / \mathrm{s}^{2}} \\
m_{C} & =0.03 \mathrm{~kg}
\end{aligned}
$$

Again look at the sub-system of $\mathrm{m}_{\mathrm{B}} \mathrm{m}_{\mathrm{C}}$, and consider the center of rotation at the edge of $\mathrm{m}_{\mathrm{C}}$.

$$
\begin{aligned}
& \sum \tau=0 \\
&=-0.09 \cdot F_{B C}+0.11 \cdot F_{B} \\
& F_{B}==\frac{0.09 \cdot F_{B C}}{0.11}=\frac{0.09 \mathrm{~m} \times 1.47 \mathrm{~N}}{0.11 \mathrm{~m}}=1.2 \mathrm{~N} \\
& \quad m_{B}=\frac{1,2 \mathrm{~N}}{9.8 \mathrm{~m} / \mathrm{s}^{2}}=0.12 \mathrm{~kg}
\end{aligned}
$$

Therefore: $F=F_{A}+F_{B C}=(0.5 \times 9,8+1,47) N=6.37 N$
====================================================
2. Two identical twin babies are born from one mother. The babies are the result of the fertilization of:
A. One ovum with two sperms
B. Two ovums with one sperm
C. One ovum with one sperm
D. Two ovums with two sperms

## Answer: C

$\rightarrow$ The principle of fertilization: one ovum is fertilized by one sperm. So the answer of $(A) \&(B)$ are wrong. The answer of $D$ probably is true but it cannot be accepted since the mother only delivers two identical babies one to each other.
3. The voltage in household wiring ( 220 volt) is used for lighting of a 100 W -bulb. The resistance R of the tungsten at $20^{\circ} \mathrm{C}$ is $89.5 \Omega$. If the temperature coefficient of tungsten $\alpha=0.0045{ }^{\circ} \mathrm{C}^{-1}$, estimate the temperature of the tungsten used as a wire resistance in the bulb.
A. $1120{ }^{\circ} \mathrm{C}$
B. $1020{ }^{\circ} \mathrm{C}$
C. $\quad 1000{ }^{\circ} \mathrm{C}$
D. $\quad 980^{\circ} \mathrm{C}$

Answer: C
Solution
$\mathrm{I}=\mathrm{P} / \mathrm{V}=100 \mathrm{~W} / 220 \mathrm{~V}=0.4545 \mathrm{~A}, \mathrm{R}(\mathrm{t})=\mathrm{V} / \mathrm{I}=484.05 \Omega$,
$\mathrm{R}(\mathrm{t})=\mathrm{Ro}\left[1+\alpha\left(\mathrm{T}-\mathrm{T}_{\mathrm{o}}\right)\right]=89.5 \Omega\left[1+0.0045{ }^{\circ} \mathrm{C}^{-1}(\mathrm{~T}-20){ }^{\circ} \mathrm{C}\right]$
$\mathrm{T}=1000{ }^{\circ} \mathrm{C}$

4. Select the statement that is not an appropriate method for separating and purifying substances:
A. Petroleum is separated from crude oil by fractional distillation
B. The mixture of various compounds could be separated by chromatography
C. Sodium Chloride is separated from the seawater by extraction
D. Iodine contained in sand mixture is separated by sublimation

Answer: C
Sodium Chloride in seawater cannot be separated by extraction but crystallization.
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5. A student sees the top and the bottom edges of a pool simultaneously at an angle of $14^{\circ}$ above the horizontal as shown in the Figure.


What is the new view angle, if he wants to see the top edge and the bottom center of the pool ( $\mathrm{n}=$ index of refraction, $\mathrm{n}_{\text {water }}=\mathrm{n}_{2}=1.33$ and $\mathrm{n}_{\text {air }}=\mathrm{n}_{1}=1$ )?
A. $\quad 28.4^{0}$
B. $\quad 38.0^{\circ}$
C. $46.8^{\circ}$
D. $51.3^{\circ}$

## Answer: D

## Solution

In order to see the bottom edge of the pool

$$
\begin{aligned}
& \mathrm{n}_{1} \sin \left(90^{\circ}-14^{\circ}\right)=\mathrm{n}_{2} \sin \theta \\
& \sin 76^{\circ}=1.33 \sin \theta \rightarrow \theta=46.85 \\
& \quad \tan \theta=1.07
\end{aligned}
$$

In order to see the bottom center of the pool

$$
\begin{aligned}
\sin \left(90^{\circ}-\theta_{2}\right) & =\mathrm{n} 2 \cdot \frac{\frac{1}{2} x}{\left(\frac{1}{4} x^{2}+h^{2}\right)}=\mathrm{n} 2 \cdot \sin \left(\left(\tan ^{-1} \frac{1}{2} \tan \theta\right)\right) \\
& =1.33 \sin \left(\tan ^{-1} 1.07 / 2\right) \\
& =0.626
\end{aligned} \quad \begin{aligned}
& \theta_{2}=90^{\circ}-\sin ^{-1}[0.626]=51.3^{\circ}
\end{aligned}
$$

6. The mechanisms of antibiotics are to inhibit the following processes, except:
A. nucleic acid synthesis
B. protein synthesis
C. capsule synthesis
D. cell wall synthesis

## Answer: C

The target of antibiotic in the body or cell of organism are: the cell wall, the membrane cell, protein synthesis and nucleic acid synthesis
7. Several indicators are used to determine the pH of river water sample. If an indicator was added to the sample, the color of indicators added to the water sample is shown below:

| Indicator added | Color of indicator in <br> water sample |
| :---: | :---: |
| Methyl orange | yellow |
| Methyl red | yellow |
| Bromthymol blue | blue |
| Phenolphthalein | colorless |

Table: pH range of indicators used:

| Indicator | Range of $\mathbf{p H}$ | Changes in color |
| :--- | :--- | :--- |
| Methyl orange | $3.1-4.4$ | red to yellow |
| Methyl red | $4.2-6.2$ | red to yellow |
| Bromthymol blue | $6.0-7.6$ | yellow to blue |
| Phenolphthalein | $8.3-9.6$ | colorless to red |

By using pH range of above indicators, the pH range of the river water is:
A. $3.1<\mathrm{pH}<7.0$
B. $4.4<\mathrm{pH}<7.6$
C. $\quad 6.0<\mathrm{pH}<8.3$
D. $7.6<\mathrm{pH}<8.3$

## Answer: D

Treatment the sample with indicators gives results:

- Methyl orange changes color to be yellow indicates that $\mathrm{pH}>4.4$
- Methyl red changes color to be yellow, indicates that $\mathrm{pH}>6.2$
- Bromthymol blue changes color to be blue, indicates that $\mathrm{pH}>7.6$
- Phenolphthalein changes color to be colorless, indicates that $\mathrm{pH}<8.3$

Therefore the pH range of the sample is $7.6<\mathrm{pH}<8.3$
8. Choose the incorrect match between organ, sensory cell and type of receptor from the table below.

| Organ | Sensory cells | Type of Receptor |
| :--- | :--- | :--- |
| I. Tongue | 1. Cone cells | a. Chemoreceptor |
| II. Ear | 2. Chemoreceptor cells | b. Photoreceptor |
| III. Nose | 3. Hair cells | c. Mechanoreceptor |
| IV. Eye | 4. Taste bud |  |

A. I, 4, a
B. II, 3, c
C. III, 2, c
D. IV, $1, \mathrm{~b}$

Answer: $\mathrm{C} \longrightarrow$ Clear
9. A woman who has four sisters married to a man who has three brothers and one sister. What is the usual probability of having a son if they have a child?
A. $12.5 \%$
B. $25 \%$
C. $50 \%$
D. $75 \%$

Answer: C. The probability to get a boy or a girl is always fifty- fifty

10. In a chemical reaction, when calcium changes (atomic number, $Z=20$ ) to form calcium ions, the ions react with carbonate ions. In this reaction each calcium atom:
A. releases one electron
B. releases two electrons
C. gains two electrons
D. increases atomic number by two

## Answer: B

${ }_{20} \mathrm{Ca} \longrightarrow{ }_{20} \mathrm{Ca}^{+2}+2 \mathrm{e}^{-}$
${ }_{20} \mathrm{Ca}$ Calcium atom has 20 electrons
${ }_{20} \mathrm{Ca}^{+2}$ Calcium ion has 18 electrons
11. X is a white solid substance. When X is heated, it produces a white solid substance Y and gas Z . The produced gas in the reaction is similar to the gas produced by burning carbon in excess of oxygen, and $Y$ is an oxide. From this information, it can be concluded that:
A. $\mathrm{X}, \mathrm{Y}$ and Z are compounds
B. Only X and gas Z are compounds
C. Y is an element and gas Z is a compound
D. X and Y are pure compounds

## Answer: A

The reaction: $\mathrm{X}(\mathrm{s}) \longrightarrow \mathrm{Y}(\mathrm{s})+\mathrm{Z}(\mathrm{g})$
Z is $\mathrm{CO}_{2}(\mathrm{~g})$ which produced from oxidation of carbon:

$$
\mathrm{C}(\mathrm{~s})+\mathrm{O}_{2}(\mathrm{~g}) \longrightarrow \mathrm{CO}_{2}(\mathrm{~g})
$$

Substance X is a carbonate salt, such as $\mathrm{MCO}_{3}$ (for example $\mathrm{M}=\mathrm{Ca}, \mathrm{Mg}, \mathrm{Ba}$ )
(a white solid) substance X is $\mathrm{MCO}_{3}$ and if $\mathrm{MCO}_{3}$ is heated, the metal oxide (white solid) and $\mathrm{CO}_{2}$ (carbon dioxide) gas will be formed.
$\mathrm{MCO}_{3}(\mathrm{~s}) \longrightarrow \mathrm{MO}(\mathrm{s})+\mathrm{CO}_{2}(\mathrm{~g})$
White white colorless

Hypertension is one of diseases that can cause death. The disease is indicated by a high blood pressure (above normal, higher than $140 / 90 \mathrm{~mm} \mathrm{Hg}$ ). The term blood pressure usually refers to the force pushing against an arterial wall. Hypertension can increase the risk of heart attacks, heart disease, strokes and kidney failure. Hypertension might be related to increasing of sodium ion concentration [atomic mass $(A)$ of sodium $=23$; atomic number $(Z)=11$ ]. Diet plays important role in hypertension, additional food such as orange, banana and vegetables could reduce blood pressure. Based on the study, orange, banana and vegetables contain potassium ion, $\mathrm{K}(\mathrm{A}=39, \mathrm{Z}=19$ ). Fifteen out of twenty people that take those diets have reduction of blood pressure (diastolic \& systolic) with obvious reduced of diastolic up to 2.4 mm Hg.
12. Active metal ion that present in orange, banana and vegetable contains
electrons and protons
A. $\quad 10$ and 11
B. 11 and 11
C. 18 and 19
D. 19 and 19

## $\underline{\text { Answer }=C}$

## Solution

The active metal ion is $\mathrm{K}^{+}$, which contains 18 electrons and 19 protons
13. Based on the above research, hypertension and the kidney failure might indicate
A. imbalance of $\mathrm{Na}^{+} / \mathrm{K}^{+}$
B. failure of $\mathrm{Na}^{+} / \mathrm{K}^{+}$re-absorption
C. substitution of $\mathrm{Na}^{+}$by $\mathrm{K}^{+}$
D. retention of $\mathrm{K}^{+}$or $\mathrm{Na}^{+}$

## Answer = A <br> \section*{Solution}

Hypertension is due to excess of $\mathrm{Na}^{+}$which can be balanced by $\mathrm{K}^{+}$.

For Test Problem No: 14 and 15

## Blood Circulatory System

The following curves show the pressure and velocity variation in the blood as it moves through the systemic circulatory system for a normal adult human. From the aorta the blood flows into the major arteries, then into the smaller arteries (arterioles), and finally into the capillaries. At each stage, these blood vessels are divided into many smaller vessels. The fluid flow (Q, volume/unit time) equal to pressure difference divided by $\mathrm{R}(\mathbf{Q}=\Delta \mathbf{P} / \mathbf{R})$, where R is the fluid resistance of a single vessel. This equation is also valid for a complex network of interconnected vessels, such as the blood vessels in the circulatory system when R is taken to be the total resistance of the network.


14．If the radius（r）of the aorta is about 0.9 cm ，use the data from the figure to estimate the flow rate of the blood Q：

A．$\quad 1.3 \times 10^{-4} \mathrm{~m}^{3} / \mathrm{s}$
B． $2.8 \times 10^{-4} \mathrm{~m}^{3} / \mathrm{s}$
C．$\quad 1.2 \times 10^{4} \mathrm{~m}^{3} / \mathrm{s}$
D．$\quad 1.3 \times 10^{4} \mathrm{~m}^{3} / \mathrm{s}$

## Answer＝A

## Solution

The flow rate $\mathrm{Q}=\mathrm{vA}=\mathrm{v} \pi \mathrm{r}^{2}=0.5 \mathrm{~m} / \mathrm{s} . \pi\left(9 \times 10^{-3} \mathrm{~m}\right)^{2}=1.3 \times 10^{-4} \mathrm{~m}^{3} / \mathrm{s}$
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15．Assume $\mathrm{Q}=1.0 \times 10^{-4} \mathrm{~m}^{3} / \mathrm{s}$ ，then the total fluid resistance R of all the arteries，arterioles，and capillaries in the body is：（density of mercury $=13,600 \mathrm{~kg} / \mathrm{m}^{3}$ ）

A．$\quad 1.1 \times 10^{-7} \mathrm{~kg} \mathrm{~m}^{-4} \mathrm{~s}^{-1}$
B． $15 \mathrm{~kg} \mathrm{~m}^{-4} \mathrm{~s}^{-1}$
C．$\quad 1.2 \times 10^{4} \mathrm{~kg} \mathrm{~m}^{-4} \mathrm{~s}^{-1}$
D． $1.1 \times 10^{8} \mathrm{~kg} \mathrm{~m}^{-4} \mathrm{~s}^{-1}$

## $\underline{\text { Answer }=\mathbf{D}}$

## Solution

From the figure above, if $\mathrm{v}=0.5 \mathrm{~m} / \mathrm{s}$, the pressure difference is about $80 \mathrm{mmHg}=0.08 \mathrm{~m} \mathrm{x} 13600$ $\mathrm{kg} / \mathrm{m}^{3} \times 9.8 \mathrm{~m} / \mathrm{s}^{2}=10,662 \mathrm{~Pa}$
$\mathrm{Q}=\Delta \mathrm{P} / \mathrm{R} \mathrm{m}^{3} / \mathrm{s} \rightarrow \mathrm{R}=\Delta \mathrm{P} / \mathrm{Q}=10,662 \mathrm{Nm}^{-2} / 1.0 \times 10^{-4} \mathrm{~m}^{3} / \mathrm{s}=1.1 \times 10^{8} \mathrm{~kg} \mathrm{~m}^{-4} \mathrm{~s}^{-1}$

For Test Problems No: 16, 17, 18

## Population Growth Pattern of Bacteria

Population growth pattern of bacteria is very important in a fermentation industry. If an Erlenmeyer flask containing nutrient broth was inoculated by a certain amount of bacterial cells $\left(\mathrm{N}_{\mathrm{o}}\right)$ at $\mathrm{T}_{0}$, after a certain period of time the bacteria will show a specific growth curve. In the beginning of growth ( $\mathrm{T}_{\mathrm{o}}-\mathrm{T}_{1}$ ) the cells are under adaptation phase. After adaptation phase, cells start to multiply by binary fission process. Each cell will divide, and become two cells. In a closed system where the nutrient supply is limited, the population of dividing cells will become relatively the same amount with the dead cells $\left(\mathrm{T}_{2}-\mathrm{T}_{3}\right)$. After $\mathrm{T}_{3}$ the amount of dead cells will be higher than the living ones, then eventually all cells will die.
16. The growth curve ( N in logarithmic scale) from the above description is:


Answer: A
Clear
17. Suppose $2 \times 10^{2}$ cells are inoculated $\left(T_{0}\right)$ and the generation time $\left(T_{g}\right)$ is 30 minutes, calculate the amount of the cells after 5 hours (ignore the adaptation time):
A. 2000
B. 4000
C. $2.05 \times 10^{5}$
D. $\quad 1.02 \times 10^{23}$

## Answer: C

Solution:
Each cell will divide, and become two cells continuously, so it can be formulated by $\mathrm{N}=\mathrm{N}_{\mathrm{o}} 2^{\mathrm{n}}$
where $\mathrm{n}=\mathrm{T} / \mathrm{T}_{\mathrm{g}}$
$\mathrm{N}=200 \times 2^{(5 / 0.5)}=2.05 \times 10^{5}$
18. The relation of cells population as a function of time can be depicted in the figure below:
$\left(1.0 \mathrm{E}+06=10^{6}\right)$


Based on the figure above, the generation time of the cells is approximately:
A. 10 minutes
B. 25 minutes
C. 35 minutes
D. 40 minutes

## Answer: B

## Solution

By using the equation $\mathrm{N}=\mathrm{N}_{\mathrm{o}} 2^{\mathrm{n}}$, and $\mathrm{n}=\mathrm{T} / \mathrm{T}_{\mathrm{g}}$, the equation becomes:
$\log \mathrm{N}=\log \mathrm{N}_{\mathrm{o}}+\mathrm{T} / \mathrm{T}_{\mathrm{g}} \log 2$ or $\log N=\log N_{o}+\frac{\log 2}{T_{g}} t$, gradient curve is $\frac{\log 2}{T_{g}}$. Taking any two points the gradient of the curve can be calculated.
Example:
Point 1: $\mathrm{N}_{1}=200, \mathrm{~T}_{1}=0$ minutes
Point 2: $\mathrm{N}_{2}=2 \times 10^{5}, \mathrm{~T}_{2}=250$ minutes
Mean $\Delta \log \mathrm{N}_{1}=3$ dan $\Delta \mathrm{T}=250$ minutes
Gradient curve $=3 / 250$ minutes $^{-1}=0.012$ mins $^{-1} \rightarrow \mathrm{~T}_{\mathrm{g}}=(\log 2) / 0.012 \sim 25$ minutes
For Test Problems No. 19, 20

## Photosynthesis

Plants need water to produce food by photosynthesis process. The water is transported through specific structures with a speed of $75 \mathrm{~cm} / \mathrm{min}$. The plant efficiency in using the water can be determined through the ratio of transpiration to photosynthesis. The ratio can be calculated by the loss of water in every gram of $\mathrm{CO}_{2}$ used in assimilation. The common ratio is 1:600. During the photosynthesis, carbon-dioxide diffuses into the leaf tissue and oxygen produced diffuses from stomata in the leaf.
19. Based on the description above, the reaction that occurred is:
A. $6 \mathrm{CO}_{2}+6 \mathrm{H}_{2} \mathrm{O}+$ Energy
$\longrightarrow$
$\mathrm{C}_{6} \mathrm{H}_{12} \mathrm{O}_{6}+6 \mathrm{O}_{2}$
B. $6 \mathrm{CO}_{2}+6 \mathrm{H}_{2} \mathrm{O}+$ Energy
$\longrightarrow$
$\mathrm{C}_{6} \mathrm{H}_{11} \mathrm{O}_{6}+6 \mathrm{O}_{2}+\frac{1}{2} \mathrm{H}_{2}$
C. $2 \mathrm{CO}_{2}+3 \mathrm{H}_{2} \mathrm{O}+$ Energy
$\longrightarrow$
$\mathrm{C}_{2} \mathrm{H}_{5} \mathrm{OH}+3 \mathrm{O}_{2}$
D. $6 \mathrm{CO}_{2}+3 \mathrm{H}_{2} \mathrm{O}+$ Energy
$\longrightarrow$
$\mathrm{C}_{6} \mathrm{H}_{6}+\frac{15}{2} \mathrm{O}_{2}$

Answer: A $\longrightarrow$ clear
20. How many liters of $\mathrm{CO}_{2}$ (at $\mathrm{T}=0^{\circ} \mathrm{C}, \mathrm{P}=1 \mathrm{~atm}$ ) are required for transpiration of 600 g water?
A. 373 L
B. 747 L
C. 1467 L
D. 1494 L

## Answer: B

## Solution :

$6 \mathrm{CO}_{2}+6 \mathrm{H}_{2} \mathrm{O} \rightarrow \mathrm{C}_{6} \mathrm{H}_{12} \mathrm{O}_{6}+6 \mathrm{O}_{2}+\mathrm{E}$
mole of $\mathrm{H}_{2} \mathrm{O}=$ mole of $\mathrm{CO}_{2}=600: 18=33.33 \mathrm{~mol}$
Volume of $\mathrm{CO}_{2}=33.33 \times 22.4 \mathrm{~L}=747 \mathrm{~L}$.


In a perfect isolated system, two bulbs are connected with a valve (see Figure above). Both bulbs are filled with a fresh air. When the valve is in closed position, the air in the first bulb has a pressure $\mathrm{P}_{1}$, volume $\mathrm{V}_{1}$ and temperature $\mathrm{T}_{1}$, and the air in the second bulb has a pressure $\mathrm{P}_{2}$, volume $\mathrm{V}_{2}$, and temperature $\mathrm{T}_{2}$. The temperature of $\mathrm{T}_{1}=\mathrm{T}_{2}$ and $\mathrm{V}_{2}=2.8 \mathrm{~V}_{1}$.

21 What is the final pressure ( P ) of the system if the valve is opened (assume the air in the bulb is an ideal gas)?
A. $\frac{P_{1}+2.8 P_{2}}{3.8}$
B. $\frac{2.8 P_{1}+P_{2}}{3.8}$
C. $\frac{P_{1}+0.8 P_{2}}{0.8}$
D. $\frac{3.8 P_{1}+P_{2}}{2.8}$

## Answer: A

## Solution

For $\mathrm{T}=$ contants, we have:

$$
\begin{aligned}
& \quad \mathrm{P}\left(\mathrm{~V}_{1}+2.8 \mathrm{~V}_{1}\right)=\mathrm{P}_{1} \mathrm{~V}_{1}+\mathrm{P}_{2} \cdot 2.8 \mathrm{~V}_{1} \\
& P=\frac{\left(P_{1}+2.8 P_{2}\right)\left(V_{1}\right)}{3.8\left(V_{1}\right)} \\
& \text { A. } P=\frac{P_{1}+2.8 P_{2}}{3.8}
\end{aligned}
$$

22. If the first bulb is filled by CO gas with pressure of 2 atm and the second bulb is filled by $\mathrm{O}_{2}$ with pressure of 1 atm . The valve is opened, the gas CO in the first bulb and gas $\mathrm{O}_{2}$ in the second bulb would mix and react completely according to the following reaction:

$$
2 \mathrm{CO}_{(\mathrm{g})}+\mathrm{O}_{2(\mathrm{~g})} \quad \rightarrow \quad 2 \mathrm{CO}_{2(\mathrm{~g})}
$$

After reaction take place completely, the gas in both bulbs consist of :
A. $\mathrm{CO}, \mathrm{O}_{2}$, and $\mathrm{CO}_{2}$
B. CO and $\mathrm{CO}_{2}$
C. $\mathrm{O}_{2}$ and $\mathrm{CO}_{2}$
D. only $\mathrm{CO}_{2}$

## Answer C

## Solution

5 mole CO equivalent to 2.5 mole $\mathrm{O}_{2}$
There is an excess of $\mathrm{O}_{2}$, therefore all CO reacts completely
The gas after reaction consists of : $\mathrm{CO}_{2}$ and unreacted $\mathrm{O}_{2}$,
$============================================================$
23. Mars' period (the time needed for one revolution about the Sun) is 684 days (Earth-days). Find the force on the Mars $\left(\mathrm{m}_{\mathrm{M}}=6.59 \times 10^{23} \mathrm{~kg}\right)$ due to the gravitational attraction of the Sun ( $\mathrm{m}_{\mathrm{S}}=1.99 \mathrm{x}$ $10^{30} \mathrm{~kg}$ ), if the distance of the Earth from the Sun is $1.50 \times 10^{11} \mathrm{~m}$. The universal gravitation constant G is $6.67 \times 10^{-11} \mathrm{Nm}^{2} / \mathrm{kg}^{2}$.
A. $5.82 \times 10^{20} \mathrm{~N}$
B. $1.09 \times 10^{21} \mathrm{~N}$
C. $1.68 \times 10^{21} \mathrm{~N}$
D. $8.96 \times 10^{21} \mathrm{~N}$

## Answer: C

## Solution:

$$
\text { From Kepler's third law: } \begin{aligned}
\frac{r_{\text {mars-sun }}}{r_{\text {earh-sun }}} & =\left(\frac{T_{m}}{T_{E}}\right)^{2 / 3} \\
& =\left(\frac{687}{365}\right)^{2 / 3} \\
& =1,52
\end{aligned}
$$

$$
\text { so } \quad \begin{aligned}
r_{\text {mars-sun }} & =1.52 \times r_{\text {earth-sun }}=1.52 \times 1.5 \times 10^{11} \mathrm{~m} \\
& =2,28 \times 10^{11} \mathrm{~m}
\end{aligned}
$$

The force on the Mars due to the Sun is :

$$
\begin{aligned}
F_{m s}=G \frac{m_{m} m_{s}}{r_{m s}^{2}} & =\frac{\left(6.67 \times 10^{-11} \mathrm{~N} . \mathrm{m}^{2} / \mathrm{kg}^{2}\right) \cdot m_{m} \cdot m_{s}}{r_{m s}^{2}} \\
& =\frac{8.73 \times 10^{43} \mathrm{Nm}^{2}}{\left(2.28 \times 10^{11} \mathrm{~m}\right)^{2}} \\
& =1.68 \times 10^{21} \mathrm{~N}
\end{aligned}
$$

24. A baby boy was born with a hemophilia disease. What is the possibility of his parents' gene related to this disease?
A. The mother is hemophilia, the father is normal
B. The mother and father are both hemophilia
C. The mother carries hemophilia gene
D. The father carries hemophilia gene

## Answer: C.

The mother is a carrier of hemophilia gene but she is not hemophilia (hemophilia is lethal for women if it is homozygote)
25. Look at the picture carefully! The secondary and the tertiary consumers are:

A. Frog, snake
B. Frog, eagle
C. Snake, eagle
D. Fox, snake

## Answer: C

Primary consumer: herbivore; Secondary Consumer : carnivore 1; Tertiary consumer: consumer 2;

